What is lithium periodic table

Lithium has three protons and four neutrons in its nucleus, and three electrons in two shells. It is located in group one, period two and block s of the periodic table.

Lithium is a chemical element of the periodic table with chemical symbol Li and atomic number 3 with an atomic weight of 6.938 u and is classed as a alkali metal.

OverviewPropertiesOccurrenceHistoryChemistryProductionApplicationsPrecautionsThe alkali metals are also called the lithium family, after its leading element. Like the other alkali metals (which are sodium (Na), potassium (K), rubidium (Rb), caesium (Cs), and francium (Fr)), lithium has a single valence electron that, in the presence of solvents, is easily released to form Li . Because of this, lithium is a good conductor of heat and electricity as well as being chemically reactive, though it is the least reactive of the al...

Lithium (pronounced as LITH-ee-em) is a soft metal with a silver appearance, represented by the chemical symbol Li. It belongs to the family of alkali metals ...

Element Lithium (Li), Group 1, Atomic Number 3, s-block, Mass 6.94. Sources, ...

Lithium - Periodic Table. Lithium is a 3. chemical element in the ...

Lithium is atomic number 3 on the periodic table with element symbol Li. Here are a collection of lithium facts, including its properties, uses, ...

Lithium in Periodic table Lithium element is in group 1 and period 2 of the Periodic table. Lithium is the s-block element and it belongs to alkali metals ...

In this comprehensive guide, you"ll learn about Lithium"s unique chemical and physical properties, trends in the periodic table, isotopes, and its historical significance.

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